

FORMULATION & LYOPHILIZATION CYCLE DEVELOPMENT OF AN ADC

Antibody Drug Conjugates (ADCs) are complex molecules consisting of a monoclonal antibody (mAb), linker, and cytotoxic drug. Formulation development must focus on maintaining the stability and potency of the mAb, while maintaining the integrity of the linker. The case study that follows describes product development for phase 1 clinical manufacturing of an ADC for an external customer. The customer had performed a pre-formulation study of the naked mAb, establishing histidine as a suitable buffer and evaluating the addition of sodium chloride or sorbitol to create an isotonic solution. However, they had not performed extensive stability or pre-formulation studies using the conjugated molecule. The goal of the study was to develop a solution formulation with an option of developing a lyophilized formulation if early stability studies indicated it was needed.

THE ADC ANALYTICAL TOOLBOX

ADC product development requires a more extensive analytical tool box than other biologic products to understand the impact of formulation and process on stability (*Figure 1*).

While the methods in Figure 1 are a standard toolbox, not all ADCs behave the same, and method optimization is required. In particular, Imaging Capillary Electrophoresis (iCE), Drug Antibody Ratio (DAR) and Free-Drug methods require development and optimization for each ADC.



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Concentration <ul style="list-style-type: none"> • UV-Vis • Nano-Drop 	Aggregation <ul style="list-style-type: none"> • SEC • DLS 	Degradation <ul style="list-style-type: none"> • ICE • Peptide Mapping
Fragmentation <ul style="list-style-type: none"> • SDS-PAGE • CE-PAGE 	Particulates <ul style="list-style-type: none"> • MFI • HIAC 	Compendial <ul style="list-style-type: none"> • pH • Appearance
Drug Antibody Ratio <ul style="list-style-type: none"> • DAR-HIC 	Linker Stability <ul style="list-style-type: none"> • Free-Drug by RP-HPLC 	Lyo <ul style="list-style-type: none"> • Residual Moisture • Recon Time

Figure 1: Analytical methods for ADC development

While the methods in Figure 1 are a standard toolbox, not all ADCs behave the same, and method optimization is required. In particular, Imaging Capillary Electrophoresis (iCE), Drug Antibody Ratio (DAR) and Free-Drug methods require development and optimization for each ADC.

Reverse-phase HPLC is commonly utilized to monitor for free-drug to ensure integrity of the linkage during formulation development and stability studies, though mass spectrometry may be used when additional sensitivity is required. The criticality of monitoring for free-drug throughout the development program will differ depending on the method of conjugation utilized. In the case study, free drug was not detected throughout development even under stressed conditions.

iCE (Figure 2) is used to monitor for changes in isoform profile and is a valuable tool for the detection of charge variants, particularly deamidation.

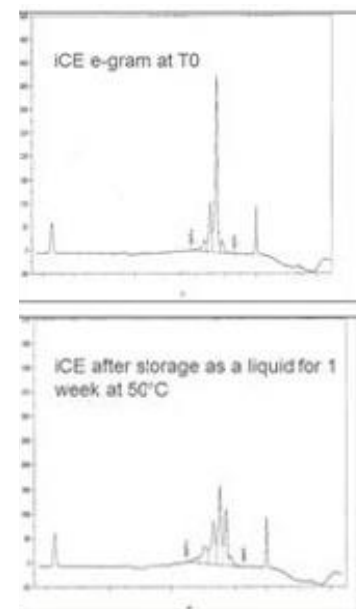


Figure 2:
iCE Electropherograms at T0 and after storage at 1 week at 50°C

DAR, a measurement of the number of cytotoxic drugs linked to each mAb, is commonly determined by hydrophobic interaction chromatography (HIC). The DAR profile differs depending on the conjugation method used. In the case study, changes in DAR (Figure 3) were not observed, even under stressed conditions.

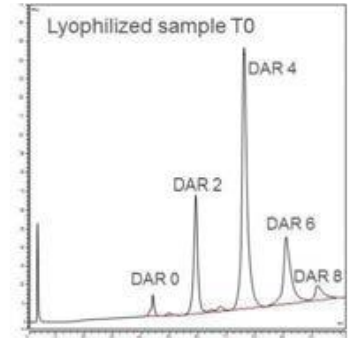


Figure 3: DAR-HIC profile for a lyophilized sample at T0

Size Exclusion Chromatography (SEC) is used to monitor for soluble aggregates. Aggregation is a common mode of degradation for ADCs, and SEC (Figure 4) was used throughout the case study.

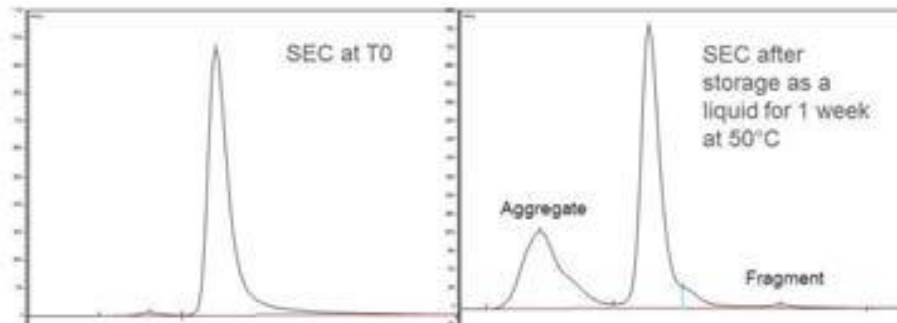


Figure 4: SEC Chromatograms at T0 and after storage as a liquid for 1 week at 50°C

SDS-PAGE (Figure 5) and CE-PAGE (Figure 6) are utilized to monitor fragmentation. Appearance of non-reduced samples will differ depending on the method of conjugation used.

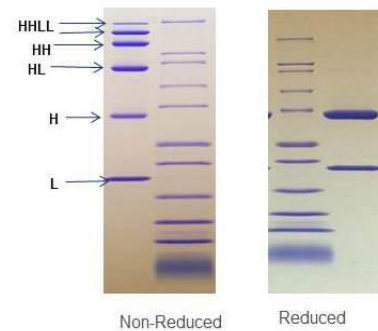


Figure 5: SDS-PAGE of non-reduced and reduced ADC

In the case study, changes in fragmentation pattern were not observed, even in stressed samples.

FORMULATION DEVELOPMENT

The customer provided the ADC as a frozen solution which was thawed and buffer exchanged as needed for formulation development studies. As the customer's preformulation work had established histidine as a suitable buffer, buffer screening studies were not repeated. Initial experiments evaluated stability with and without polysorbate 80, with and without sorbitol (225 mM), and at pH's of 5.5, 6.0, and 6.5. Liquid samples were subjected to agitation, freeze-thaw cycles, and storage at 40°C. Analysis included SEC, iCE, DAR-HIC, UV-Vis, as well as two dimensional Fourier-Transform Infrared Spectroscopy (2D-FTIR) used for the determination of secondary structural changes. The addition of sorbitol and polysorbate 80 had no effect on the stability of the ADC during agitation, free-thaw, and accelerated stability studies. The pH had the greatest impact on stability: the ADC appeared to be more stable at pH 6.0 by SEC and at pH 5.5 and 6.0 by iCE (Figure 7).

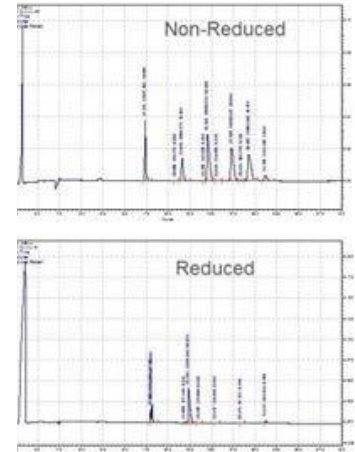
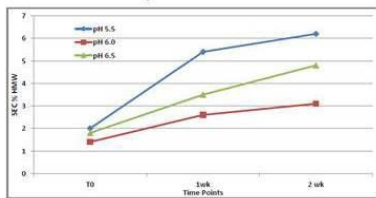
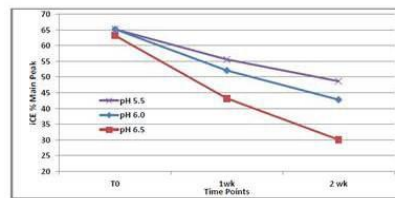


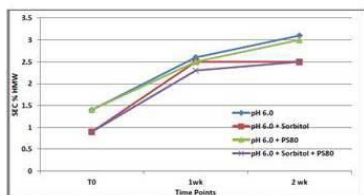
Figure 6: CE-PAGE of non-reduced and reduced ADC



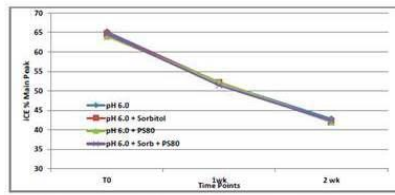
The Effect of pH on the Percent HMW for ADC as Determined by SEC When Stored at 40°C for 2 Weeks.



The Effect of pH on the Percent Main Peak for ADC as Determined by iCE When Stored at 40°C for 2 Weeks.



The Effect of Sorbitol and PS80 at pH 6.0 on the Percent HMW for ADC as Determined by SEC When Stored at 40°C for 2 Weeks.



The Effect of Sorbitol and PS80 at pH 6.0 on the Percent Main Peak for ADC as Determined by iCE When Stored at 40°C for 2 Weeks.

Figure 7: SEC and iCE analysis of liquid samples following agitation, freeze-thaw, or storage at 40°C

The initial screening study indicated susceptibility to deamidation. To mitigate risk of development delays, work was immediately begun to establish a formulation suitable for lyophilization, should it be required. Sorbitol is not typically used as a stabilizing agent during lyophilization. The longest step in a lyophilization cycle is primary drying, and increasing the temperature at which it occurs can reduce its duration. The glass transition temperature (Tg') of the formulation is an indication of the maximum product temperature for primary drying. Common non-reducing disaccharides, such as sucrose and trehalose, have a Tg' of -32°C and -30°C respectively. Sorbitol, has a low Tg' of -43°C, making it a poor choice for an efficient freeze-drying process.

Formulation	pH	Tween (%)	Sucrose (mM)	Trehalose (mM)
13	5.5	0	225	0
14	5.5	0	0	225
15	6.0	0	225	0
16	6.0	0	0	225
17	5.5	0.02	225	0
18	5.5	0.02	0	225
19	6.0	0.02	225	0
20	6.0	0.02	0	225

Table 1: Stabilizing Excipient Screening Study

Formulations containing either sucrose or trehalose were prepared and stored as solutions (2-8°C and 25°C) and freeze-dried solids (25°C and 40°C). Liquid samples were also subjected to agitation and freeze-thaw cycles. The full screening study is described in Table 1. As in the prior study with sorbitol, formulations prepared at pH 6 (Figure 8) exhibited better results on stability than formulations prepared at pH 5. No large differences in stability data were

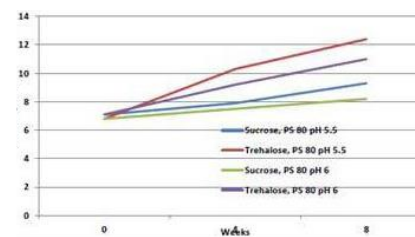


Figure 8: The Percent Basic Species as determined by iCE for samples stored at 40°C for up to 8 weeks

observed for samples prepared with sucrose or trehalose or between samples prepared with or without polysorbate 80. The formulation candidate chosen for further studies contained sucrose at pH 6 without polysorbate 80.

Based on the early stability results from the two screening studies, the customer made a decision to pursue a lyophilized formulation. While the ADC appeared to have good chemical stability in the solid state as a 30 mM histidine, 225 mM (7.7%) sucrose formulation, the cake appearance was unacceptable, showing shrinkage of the cake within the vials (Figure 9).

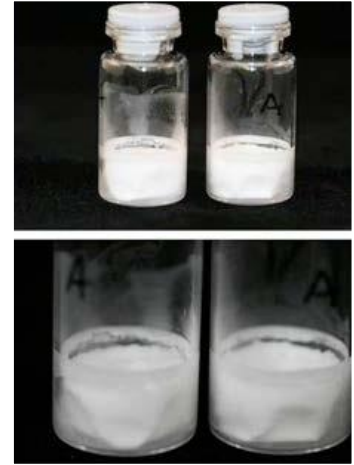


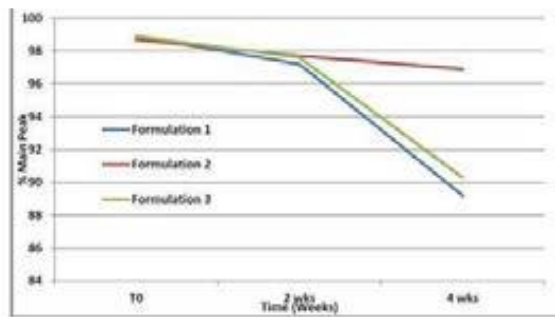
Figure 9: Photographs of lyophilized ADC-sucrose formulation showing shrinkage of cake

The addition of mannitol was investigated to improve the cake appearance. Three formulations representing differing ratios of sucrose and mannitol (Table 2 - next page), were formulated and lyophilized and placed on stability at 50°C for 4 weeks and 25°C and 40°C for 2 months.

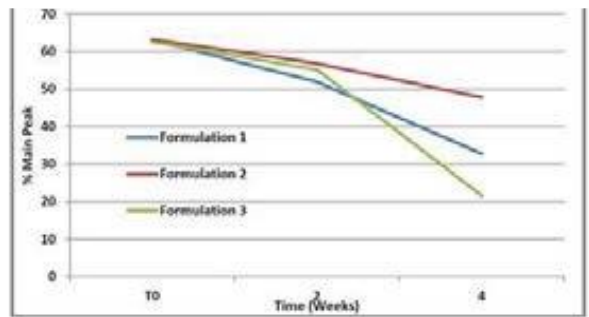
Significant differences were observed (Figure 10) by both iCE and SEC: both test methods demonstrated that the 3% sucrose – 3% mannitol combination offered the greatest stability.

Sample ID	Histidine (mM)	Sucrose	Mannitol
Formulation 1	30	225 mM (7.7%)	274 mM (5%)
Formulation 2	30	88 mM (3%)	165 mM (3%)
Formulation 3	30	160 mM (5.5%)	274 mM (5%)

Figure 10: SEC and iCE results from Sucrose-Mannitol studies



Percent Main Peak from SEC for Formulations 1, 2, and 3 after Storage at 50°C for 4 Weeks.



Percent Main Peak from iCE for Formulations 1, 2, and 3 after Storage at 50°C for 4 Weeks.

Figure 10: SEC and iCE results from Sucrose-Mannitol studies

LYOPHILIZATION DEVELOPMENT

Development of a lyophilization cycle was conducted with the 30 mM histidine, 88 mM sucrose (3%), 165 mM mannitol (3%) pH 6.0 formulation. Lyophilization cycles typically consist of three steps (Figure 11), freezing, primary drying, and secondary drying. During the freezing step, the shelf temperature is reduced and ice nucleation occurs. The primary drying step is the longest step in the process. The shelf temperature is increased and a vacuum is applied, resulting in ice sublimation and the removal of approximately 90% of the water in the vial.

During the secondary drying step, the shelf temperature is increased further and the more tightly-bound water is removed. Typically, the length of the secondary drying step controls the final moisture content of the product.

Formulations containing mannitol, or other crystallizing excipients, typically require annealing to ensure that the mannitol has fully crystallized. Lack of mannitol crystallization during the lyophilization process can result in the formation of the hydrate form of mannitol, which can crystallize at a later time, resulting in the release of water into the cake and

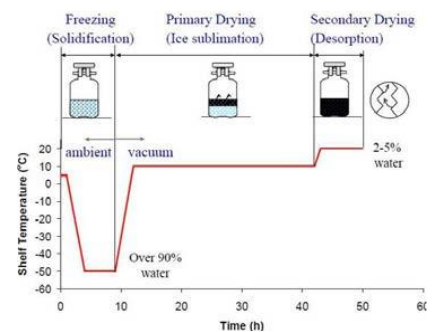


Figure 11: The three-step lyophilization process

negatively impacting stability. Annealing is a step in the lyophilization process in which the frozen product is warmed to a temperature greater than the T_g' but not great enough to result in melting. For the selected formulation, Differential Scanning Calorimetry (DSC) was used to determine that the onset of crystallization occurred at approximately 10 minutes at temperatures from -15°C to -20°C .

A conservative maximum product temperature for primary drying can be estimated from the T_g' . Drying at temperatures above the T_g' can result in the freeze-concentrated formulation undergoing viscous flow once the supporting ice structure is removed, resulting in collapse. The T_g' of formulation was determined by DSC to be between -32°C and -34°C .

Freeze-dry microscopy was used to confirm the collapse temperature (T_c) and the onset of collapse occurred at -32°C (Figure 12).

Actual conditions for primary drying were confirmed experimentally through edge of failure cycles, in which the cycle is repeated at greater primary drying temperatures until visual cake appearance is unacceptable. Cake shrinkage was first observed in cycles with primary drying temperatures of -18°C , while cycles that ensured that primary drying remained at -21°C or less during primary drying produced visually acceptable lyophilized samples.

A design space, or safe zone of operation, was constructed for the ADC product. The design space is defined by three factors: (1) the maximum sublimation rate supported by the freeze-dryer, (2) the heat transfer coefficient for the vial (K_v)

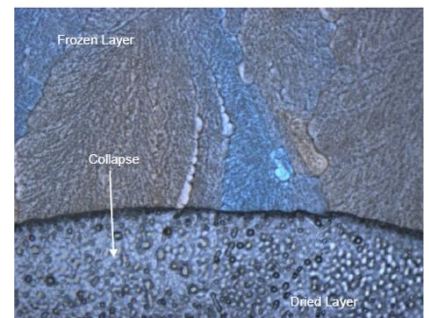


Figure 12: Freeze-Dry microscopy image at -32°C showing the onset of collapse

and (3) the resistance of the dried product layer (R_p) of the formulation. The K_v and R_p were determined using an in-process monitoring technique known as Tunable-Diode Laser Absorption Spectroscopy (TDLAS).

Equipment capability varies between lyophilizers and can be experimentally determined using ice-slab testing. The laboratory Lyostar freeze-dryers have lower capability than production dryers, and as a conservative approach, the capability line for the laboratory dryers was used for construction of the design space.

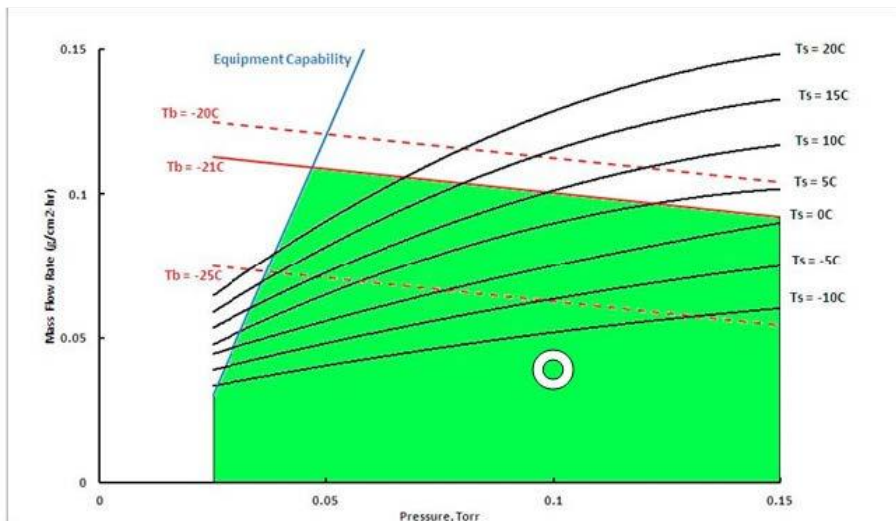


Figure 13: Design space for primary drying of the ADC product

In the design space (Figure 13), the equipment capability line is blue, the shelf temperature (T_s) isotherms are black and the product temperature isotherms (T_b) are dashed red lines. The experimentally determined maximum product temperature of -21°C is marked with a solid red line. The most efficient cycle is represented by conditions in the upper left corner of the green zone and would require 23 hours. These conditions were utilized in a laboratory challenge run, utilizing an IgG in place of the ADC, and were successful in

maintaining the product temperature less than -21°C during primary drying and of producing cakes of expected acceptable appearance. The condition marked with a white circle was used for the demonstration batch and represents a cycle time of 40 hours. The demonstration batch was placed on long term stability for 18 months at 5°C and 25°C as well as for 6 months at 40°C. By all test methods, the lyophilized product remained stable.

SUMMARY

Development of a lyophilization cycle was conducted with the 30 mM histidine, 88 mM sucrose (3%), 165 mM mannitol (3%) pH 6.0 formulation. Lyophilization cycles typically consist of three steps (*Figure 11*), freezing, primary drying, and secondary drying. During the freezing step, the shelf temperature is reduced and ice nucleation occurs. The primary drying step is the longest step in the process. The shelf temperature is increased and a vacuum is applied, resulting in ice sublimation and the removal of approximately 90% of the water in the vial.

Early stability studies indicated that the ADC was not stable in a liquid formulation, but development activities were rapidly transitioned to focus on a lyophilized presentation and clinical timelines were unaffected. The use of a scientifically-sound lyophilization development process resulted in an efficient, well-defined, robust process, which resulted in a stable product. The formulation and manufacturing process utilized for the laboratory demonstration batch were successfully transferred to manufacturing scale at Baxter's Halle Germany facility for clinical manufacturing.

SOURCE

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